



# *Condensed Matter Physics*

## *Molecular dynamics II.*

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# Transition metal



Periodic table of elements showing the distribution of elements across the periodic table, color-coded by groups.

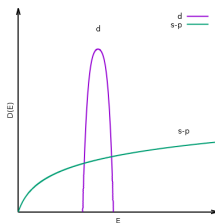
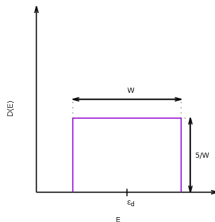


Figure: Density of states



*Figure:* Density of states in the Friedel-model

There are  $2 \cdot 5$  d elektron states

$$E_{bond} = 2 \int_{\epsilon_d}^{E_f} (E - \alpha) D(E) dE$$

For our case

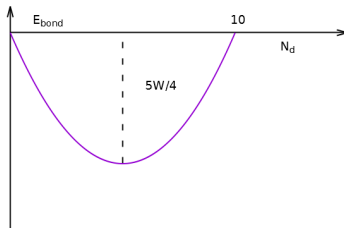
$$E_{bond} = 2 \int_{\epsilon_d - \frac{W}{2}}^{E_f} (E - \epsilon_d) \frac{5}{W} dE =$$

The Fermi energy is determined by the number of electrons

$$N_d = 2 \int_{\varepsilon_d - \frac{W}{2}}^{E_f} \frac{5}{W} dE = \frac{10}{W} \left( E_f - \varepsilon_d + \frac{W}{2} \right)$$

The bounding energy is

$$E_{bond} = -\frac{W}{20} N_d (10 - N_d)$$



## Second moment of the density of states



Let us consider a base system given by  $|i\rangle$

The density of states corresponding to the state  $|i\rangle$  can be defined as

$$d_i(E) = \sum_{\text{all } E_k} \delta(E - E_k) \langle i | \psi_k \rangle \langle \psi_k | i \rangle$$

The second moment of  $d_i(E)$  is

$$\mu_i^{(2)} = \int (E - H_{ii})^2 d_i(E) dE$$

That is

$$\mu_i^{(2)} = \int_{\text{band}} \sum_{E_k} (E - H_{ii})^2 \delta(E - E_k) \langle i | \psi_k \rangle \langle \psi_k | i \rangle dE$$

leading to

$$\begin{aligned} \mu_i^{(2)} &= \sum_{E_k} (E_k - H_{ii})^2 \langle i | \psi_k \rangle \langle \psi_k | i \rangle \\ \mu_i^{(2)} &= \langle i | \underbrace{\left( \sum_{E_k} |\psi_k\rangle (E_k - H_{ii})^2 \langle \psi_k | \right)}_{(\hat{H} - H_{ii})^2} | i \rangle \\ \mu_i^{(2)} &= \sum \langle i | (\hat{H} - H_{ii}) | i' \rangle \langle i' | (\hat{H} - H_{ii}) | i \rangle \end{aligned}$$

## Second moment of the density of states



Since

$$\langle i | \hat{H} - H_{ii} | i \rangle = 0 \qquad \langle i | \hat{H} - H_{ii} | i' \rangle = H_{ii'} \qquad i \neq i'$$

one gets

$$\mu_i^{(2)} = H_{ii'} H_{i' i}$$

That is

$$\mu_i^{(2)} = z\beta^2$$

with  $z$  is the number of core electrons, and  $\beta$  is the “hopping” probability.  
In the Friedel model

$$\mu_i^{(2)} = \int_{\varepsilon_d - \frac{W}{2}}^{\varepsilon_d + \frac{W}{2}} (E - \varepsilon_d)^2 \frac{1}{W} dE = \frac{W^2}{12}$$

From this one gets  $W$

$$W = (12z)^{\frac{1}{2}} |\beta| \qquad \beta(R) \approx N_d e^{-\kappa R}$$

$\beta$  is an exponential function with characteristic length scale  $1/\kappa$ .

Bonding energy

$$E_{bond} = -\frac{bN_d^2(12z)^{\frac{1}{2}}}{20}e^{-\kappa R}N_d(10 - N_d) \quad (1)$$

Repulsive energy

$$E_{rep} = aN_d^2e^{-2\kappa R} \quad (2)$$

Total energy

$$E_{\omega h} = E_{rep} + E_{bond} \quad (3)$$

With these

$$E_{\omega h} = Ae^{-2\kappa R} - Be^{-\kappa R} \quad (4)$$

$B$  is calculated,  $A$  and  $\kappa$  are empirical parameters.

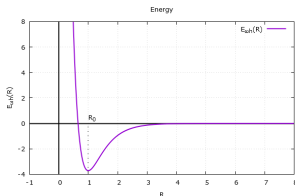
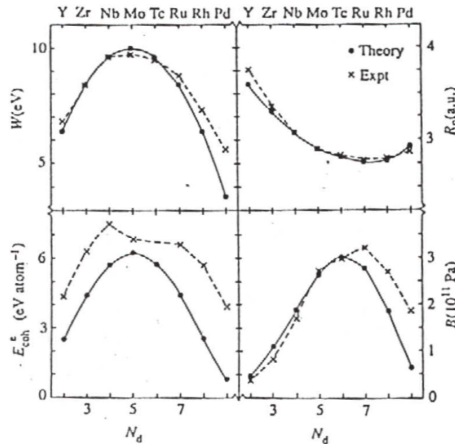


Figure:  $E_{\omega h}(R)$



**Figure:** Predicted variations in properties across the 4d transient metal series based on the Friedel model in the second moment approximation compared with experiment (from Pettifor 1987). Top left: the bandwidth,  $W$ . Top right: the equilibrium bond length,  $R_0$  in Bohr. Bottom left: the cohesive energy,  $E_{coh}^e$ . Bottom right: the bulk modulus  $B$  in  $10^{11}$  Pa units.



They assumed that the states are “localized” to the  $i$ th atom  
The bounding energy is

$$E_{bond} = 2 \sum_i \int_{-\infty}^{E_f} (E - \varepsilon_i) d_i(E) dE$$

For the density of state Finnis suggested a Gaussian function:

$$d_i(E) = \frac{5}{\sqrt{2\pi\mu_i^{(2)}}} \exp\left(-\frac{(E - \varepsilon_i)^2}{2\mu_i^{(2)}}\right)$$

So the energy of the electron “bounded to the  $i$ th atom is:

$$E_{bond}^{(i)} = 2 \int_{-\infty}^{E_f} d_i(E)(E - \varepsilon_i) dE = -\frac{10}{\sqrt{2\pi}} \sqrt{\mu_i^{(2)}} \exp\left(-\frac{(E_f - \varepsilon_i)^2}{2\mu_i^{(2)}}\right)$$

Assuming that the atoms are neutral

$$N_d = \int_{-\infty}^{E_f} d_i(E) dE$$

Let us introduce  $L_i$  with:

$$N_d = \frac{10}{\sqrt{\pi}} \int_{-\infty}^{L_i} e^{-x^2} dx \qquad L_i = \frac{(E_f - \varepsilon_i)}{\sqrt{2\mu_i^{(2)}}}$$

Assuming that the number of electrons ( $N_d$ ) are the same for all atoms one sees  $L_i$  is independent from  $i$ :

$$L_i := L$$

Az így kapott kötési energián azt látjuk, hogy nem párok összegeként áll elő:

$$E_{bond}^{(i)} = const. \sqrt{\mu_i^{(2)}}$$

Where

$$\mu_i^{(2)} = \sum_j H_{ij} H_{ji} = \sum_j \phi(r_{ij})$$

is a sum.

Total energy

$$V(\vec{r}_1, \vec{r}_2, \dots) = \sum_{i \neq j} \phi_1(\vec{r}_1 - \vec{r}_i) + \sum_i \sqrt{\sum_j \phi(r_{ij})}$$

Generalized form

$$V(\vec{r}_1, \vec{r}_2, \dots) = \sum_{i \neq j} \phi_1(\vec{r}_1 - \vec{r}_i) + \sum_i F \left( \sum_j \phi(r_{ij}) \right)$$

where  $F$  is some function.