

A6180/8180: Modern Scattering Methods in Materials Science

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Scattering by non-crystalline materials

Debye-formula

Zernicke-Prins equation

the scattered intensity in the most general form:

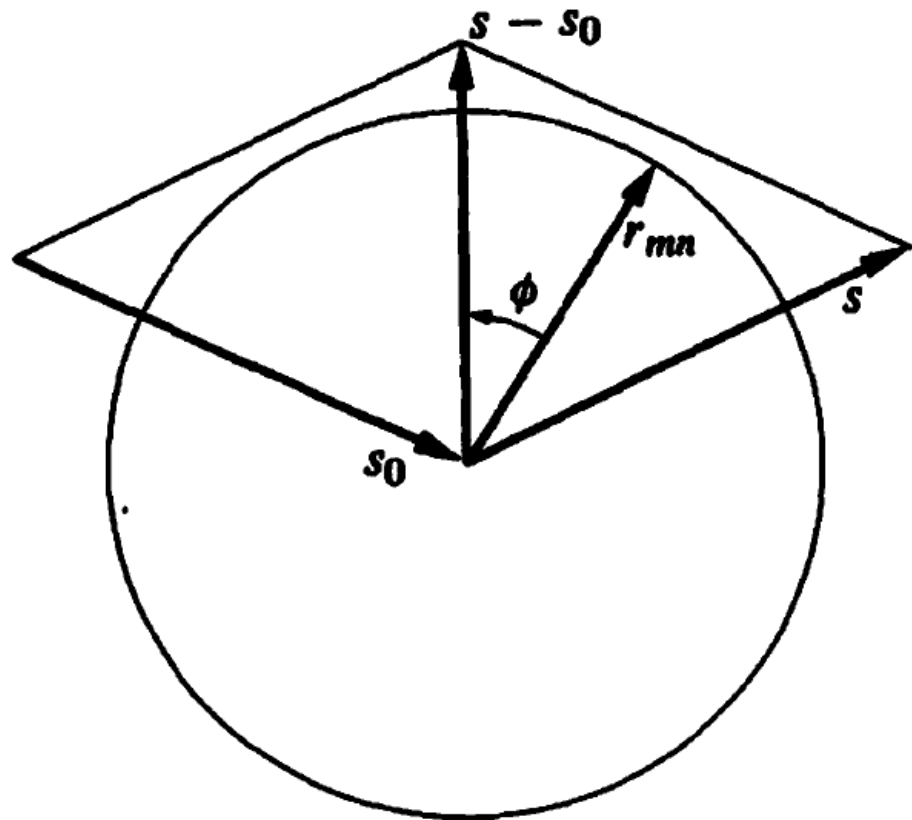
$$I_{eu} = \sum_m f_m e^{(2\pi i/\lambda)(s-s_0) \cdot \mathbf{r}_m} \sum_n f_n e^{-(2\pi i/\lambda)(s-s_0) \cdot \mathbf{r}_n}$$

where summation has to be done over all the atoms within the illuminated region,

with the difference vector $\mathbf{r}_{mn} = \mathbf{r}_m - \mathbf{r}_n$:

$$I_{eu} = \sum_m \sum_n f_m f_n e^{(2\pi i/\lambda)(s-s_0) \cdot \mathbf{r}_{mn}}$$

Fig. 10.1 Relations involved in letting the \mathbf{r}_{mn} vector take all orientations in space.



with $k = (4\pi \sin \theta)/\lambda$, the average for each exponential term:

$$\begin{aligned}\langle e^{(2\pi i/\lambda)(\mathbf{s}-\mathbf{s}_0)\cdot\mathbf{r}_{mn}} \rangle &= \\ &= \frac{1}{4\pi r_{mn}^2} \int_{\phi=0}^{\pi} e^{ikr_{mn}\cos\phi} 2\pi r_{mn}^2 \sin \phi \, d\phi \\ &= \frac{\sin kr_{mn}}{kr_{mn}}\end{aligned}$$

after summation

$$I_{eu} = \sum_m \sum_n f_m f_n \frac{\sin kr_{mn}}{kr_{mn}}$$

Debye-formula

gas of polyatomic molecule:

$$I_{eu}/N = \sum_m \sum_n f_m f_n \frac{\sin kr_{mn}}{kr_{mn}} + R \sum_m i(M)_m$$

where the last term on the right-hand side is the *modified* Compton scattering and N is the number of molecules in the illuminated volume.

carbon-tetrachloride CCl_4 :

$$\begin{aligned}
 & f_{\text{C}} \left\{ f_{\text{C}} + 4f_{\text{Cl}} \frac{\sin kr(\text{C} - \text{Cl})}{kr(\text{C} - \text{Cl})} \right\} + \\
 & + 4f_{\text{Cl}} \left\{ f_{\text{Cl}} + f_{\text{C}} \frac{\sin kr(\text{C} - \text{Cl})}{kr(\text{C} - \text{Cl})} \right\} + \\
 & + 3f_{\text{Cl}} \frac{\sin kr(\text{Cl} - \text{Cl})}{kr(\text{Cl} - \text{Cl})} \left\{ \right.
 \end{aligned}$$

For a tetrahedral molecule $r(\text{Cl} - \text{Cl}) = \sqrt{\frac{8}{3}} r(\text{C} - \text{Cl})$

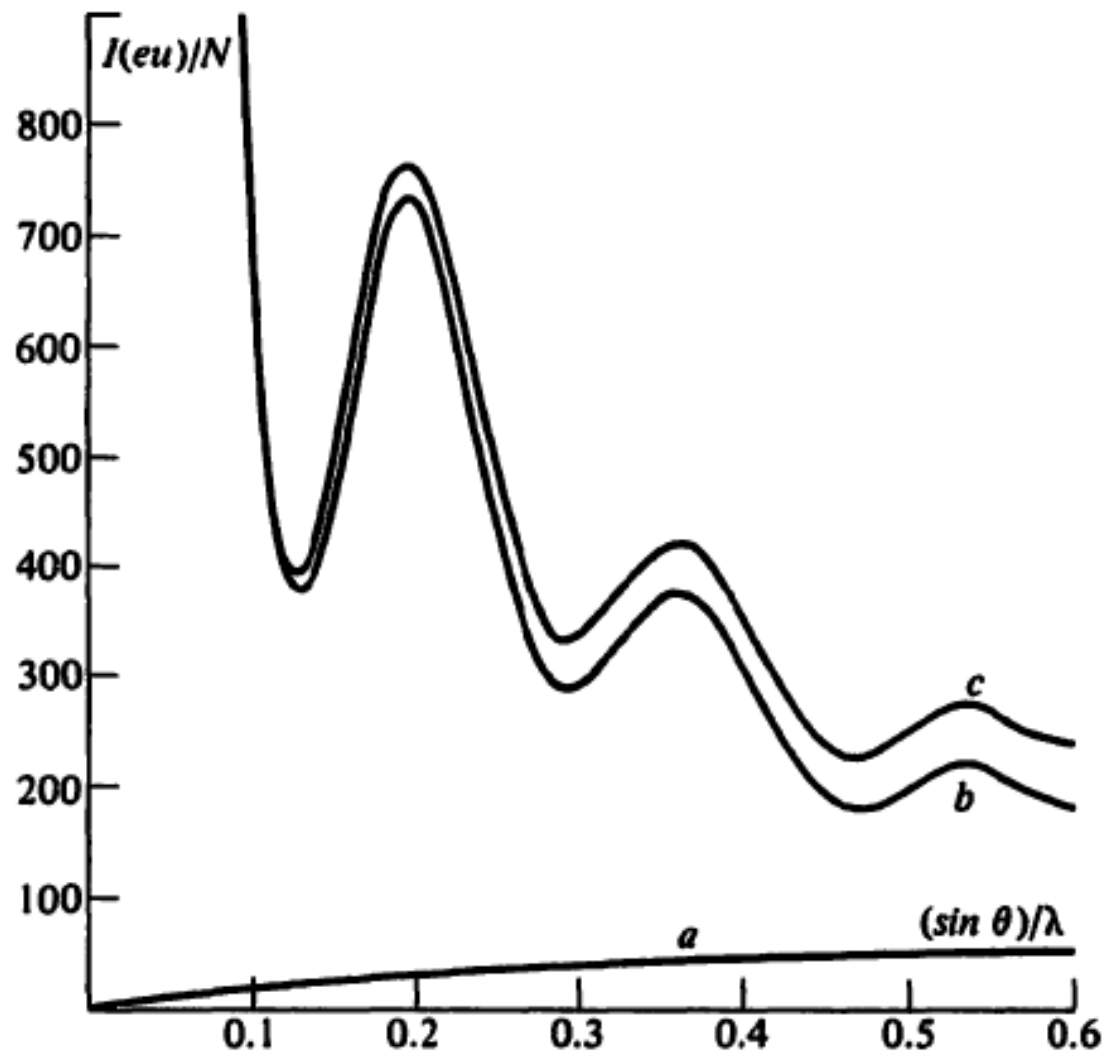
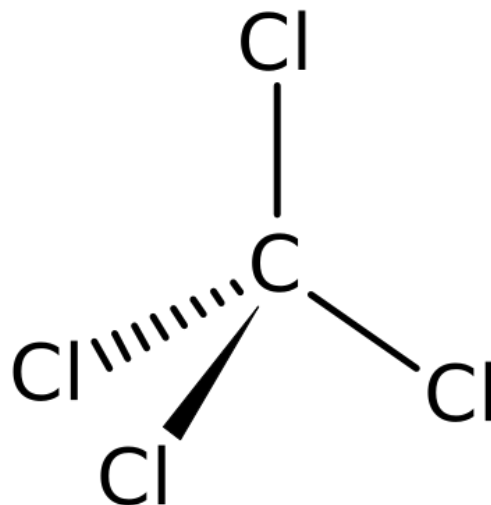
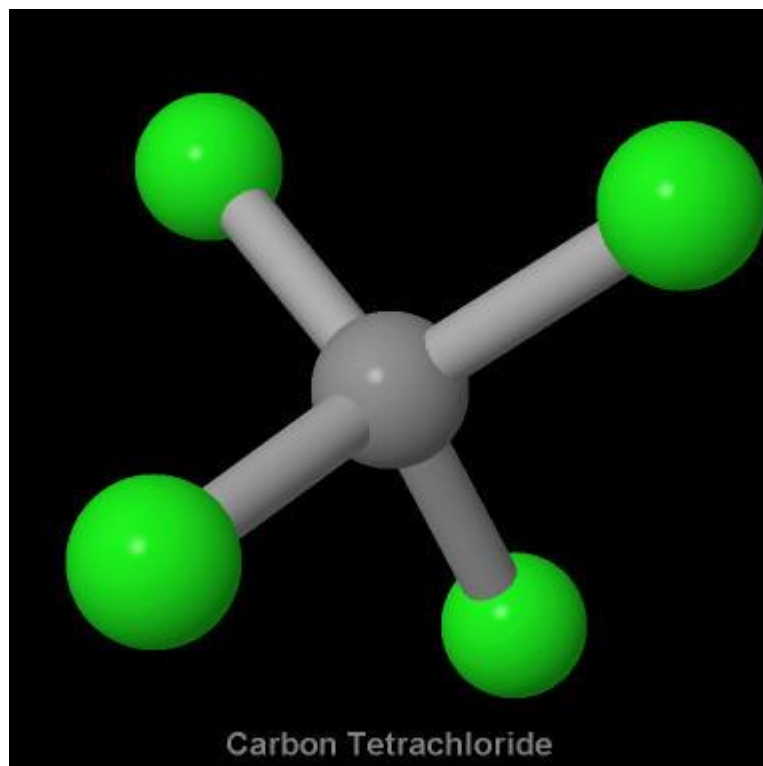


Fig. 10.2 The intensity in electron units per molecule for a CCl_4 gas in which the carbon-chlorine distance is taken as $r = 1.82 \text{ \AA}$. (a) The modified intensity per molecule. (b) The unmodified intensity. (c) The total intensity per molecule, unmodified plus modified.



$$r_{\text{C-Cl}} = 0.182 \text{ nm}$$

Scattering by liquid or non-crystalline materials

if there is only one kind of atom, the scattered intensity will be:

$$I_{eu} = \sum_m f^2 + \sum_m f^2 \sum_{n \neq m} e^{(2\pi i/\lambda)(\mathbf{s}-\mathbf{s}_0) \cdot \mathbf{r}_{nm}},$$

we introduce a density function: $\rho_m(\mathbf{r}_{nm})$

where $\rho_m(\mathbf{r}_{nm}) dV_n$ is the number of atoms in dV_n at \mathbf{r}_{nm} .

With this the summation can be replaced by the integral:

$$I_{eu} = \sum_m f^2 + \sum_m f^2 \int_S \rho_m(\mathbf{r}_{nm}) e^{(2\pi i/\lambda)(\mathbf{s}-\mathbf{s}_0) \cdot \mathbf{r}_{nm}} dV_n$$

Scattering by liquid or non-crystalline materials

with introducing the average density, ρ_a :

$$I_{eu} = \sum_m f^2 + \sum_m f^2 \int_S [\rho_m(\mathbf{r}_{nm}) - \rho_a] e^{(2\pi i/\lambda)(\mathbf{s}-\mathbf{s}_0)\cdot\mathbf{r}_{nm}} dV_n + \\ + \sum_m f^2 \int_S \rho_a e^{(2\pi i/\lambda)(\mathbf{s}-\mathbf{s}_0)\cdot\mathbf{r}_{nm}} dV_n$$

for a fixed distance $\mathbf{r}_{mn}=\mathbf{r}$ let $\rho(\mathbf{r}) = \langle \rho_m(\mathbf{r}_{nm}) \rangle$, where the average

is over all $\rho_m(\mathbf{r})$ within the sample at the distance \mathbf{r} from any atom in the sample.

Scattering by liquid or non-crystalline materials

for a fixed distance in the sample,

$$\mathbf{r}_{nm} = \mathbf{r}, \text{ let } \rho(\mathbf{r}) = \langle \rho_m(\mathbf{r}_{nm}) \rangle,$$

where the average is taken over all $\rho_m(\mathbf{r})$ within the sample at the distance \mathbf{r} from any atom.

With this, $[\rho_m(\mathbf{r}_{nm}) - \rho_a]$ in the integral can be replaced by:

$$\rho(\mathbf{r}) - \rho_a$$

Scattering by liquid or non-crystalline materials

For a mon-atomic amorphous sample the term

$$\rho(\mathbf{r}) - \rho_a$$

goes to zero at distances of \mathbf{r} larger than a few atomic distances.

Since the volume S is considerably larger than a few atomic distances the summation over m can be replaced by the number of atoms in S , i.e., N .

Scattering by liquid or non-crystalline materials

if there is no preferred orientation in the sample,

$$\rho(\mathbf{r}) - \rho_a$$

will have spherical symmetry and can be written as:

$$\rho(r) - \rho_a$$

Scattering by liquid or non-crystalline materials

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Using the notations in Fig. 10.1 and the variable $k = 4\pi \sin \theta / \lambda$ we obtain:

$$\begin{aligned} Nf^2 \int_0^\infty \int_0^\pi [\rho(r) - \rho_a] e^{ikr \cos \phi} 2\pi r^2 \sin \phi \, d\phi \, dr &= \\ &= Nf^2 \int_0^\infty 4\pi r^2 [\rho(r) - \rho_a] \frac{\sin kr}{kr} \, dr \end{aligned}$$

Scattering by liquid or non-crystalline materials

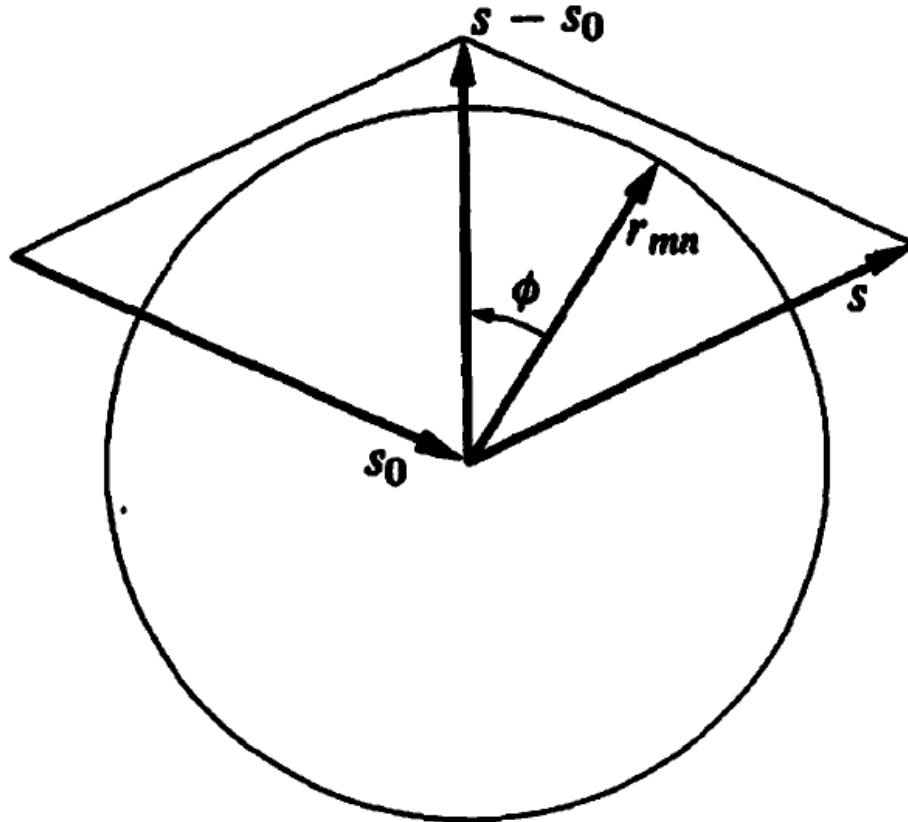


Fig. 10.1 Relations involved in letting the r_{mn} vector take all orientations in space.

Scattering by liquid or non-crystalline materials

after some rearrangements:

$$I_{eu} = Nf^2 + Nf^2 \int_0^\infty 4\pi r^2 [\rho(r) - \rho_a] \frac{\sin kr}{kr} dr + \\ + f^2 \sum_m e^{-(2\pi i/\lambda)(s-s_0) \cdot r_m} \int_S \rho_a e^{(2\pi i/\lambda)(s-s_0) \cdot r_n} dV_n.$$

next we analyze the integral in the third term:

$$\int_S \rho_a e^{(2\pi i/\lambda)(s-s_0) \cdot r_n} dV_n.$$

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since ρ_a is a constant, the integral can be written as:

$$\rho_a \int_S e^{\left(2\pi\frac{i}{\lambda}\right)(\mathbf{s}-\mathbf{s}_0)\cdot\mathbf{r}_n} dV_n$$

the integral here is the Fourier transform of the form-factor of the volume of the sample within S . If S is macroscopic then the integral is close to a delta function around the direct-beam direction, and therefore has no effect on the diffraction experiment.

The third term on the right-hand side of I_{eu} can be omitted.

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the scattered intensity per atom will become:

$$I'_{eu}/N = f^2 + f^2 \int_0^\infty 4\pi r^2 [\rho(r) - \rho_a] \frac{\sin kr}{kr} dr.$$

We introduce the abbreviation:

$$i(k) = \frac{I'_{eu}/N - f^2}{f^2},$$

and do some rearrangements:

$$ki(k) = 4\pi \int_{r=0}^\infty r[\rho(r) - \rho_a] \sin kr dr$$

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with taking into account the rules of Fourier transformation:

$$\phi(k) = 4\pi \int_0^\infty f(r) \sin kr \, dr,$$

$$f(r) = \frac{1}{2\pi^2} \int_0^\infty \phi(k) \sin rk \, dk.$$

the inverse Fourier transformation provides:

$$r[\rho(r) - \rho_a] = \frac{1}{2\pi^2} \int_0^\infty ki(k) \sin rk \, dk$$

and with some rearrangements:

Scattering by liquid or non-crystalline materials

we arrive at the Zernicke-Prins equation:

$$4\pi r^2 \rho(r) = 4\pi r^2 \rho_a + \frac{2r}{\pi} \int_0^\infty k i(k) \sin rk \, dk.$$

$4\pi r^2 \rho(r)$ is the *radial-distribution-function* (RDF)

F. ZERNICKE and J. A. PRINS, *Zeit Physik*, **41, 184 (1927).**

Scattering by liquid or non-crystalline materials

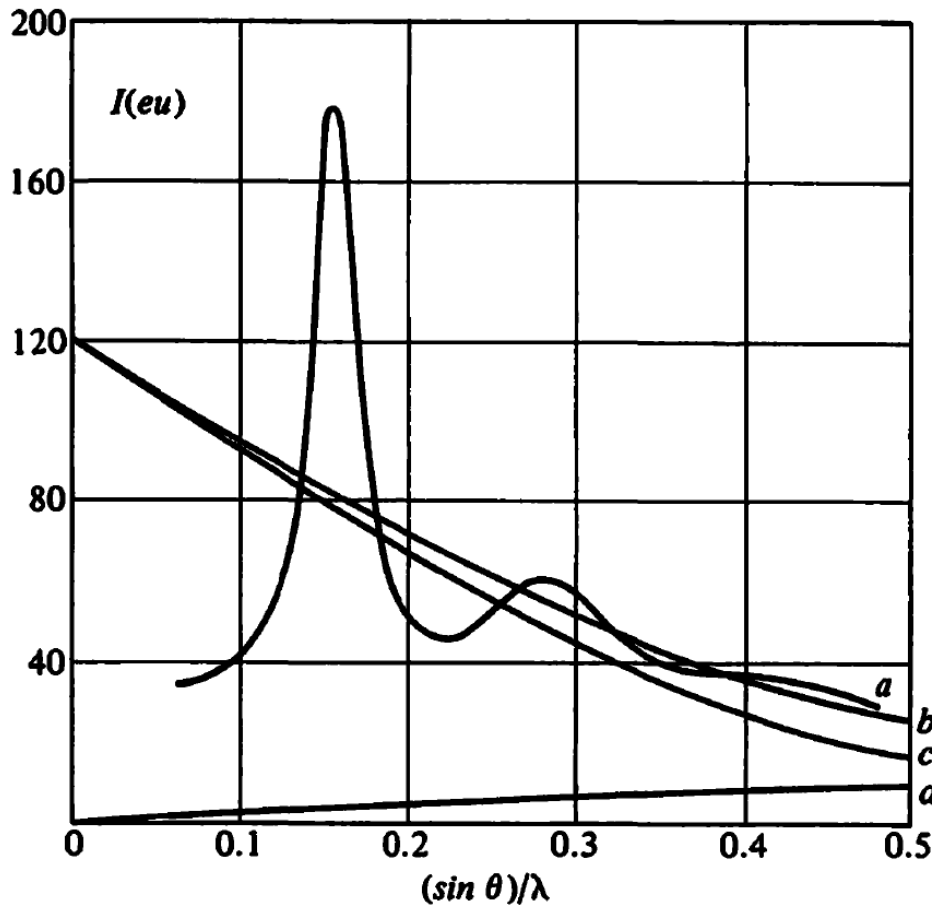


Fig. 10.4 (a) Total intensity curve for liquid sodium in electron units per atom, unmodified plus modified. (b) Total independent scattering per atom. (c) Independent unmodified scattering per atom f^2 . (d) Modified scattering per atom $i(M)$.

Scattering by liquid or non-crystalline materials

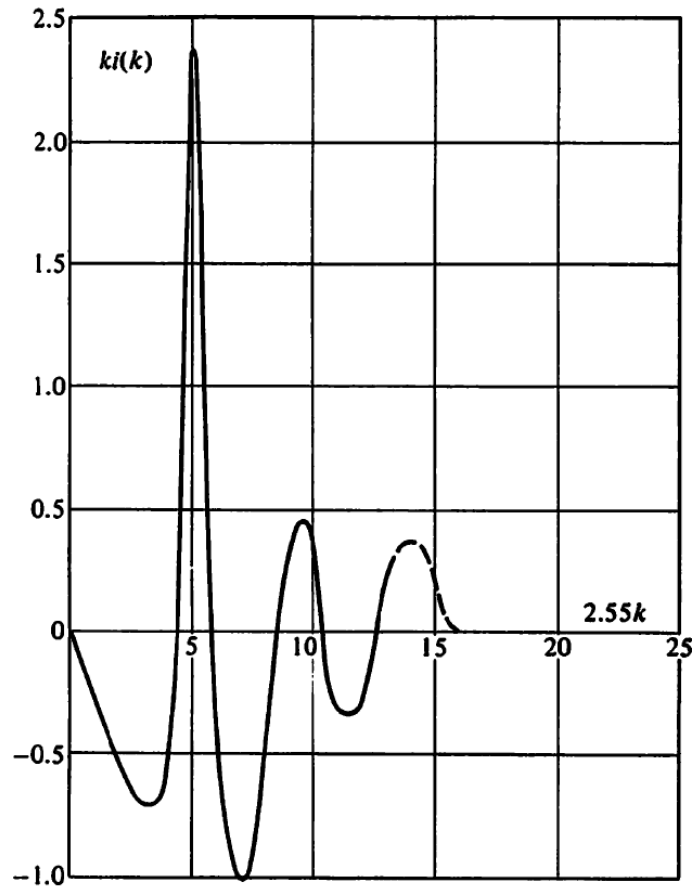


Fig. 10.5 The experimental curve $ki(k)$ for liquid sodium. The abscissa scale $2.55k$ was chosen for convenience in the use of a harmonic analyzer.

Scattering by liquid or non-crystalline materials

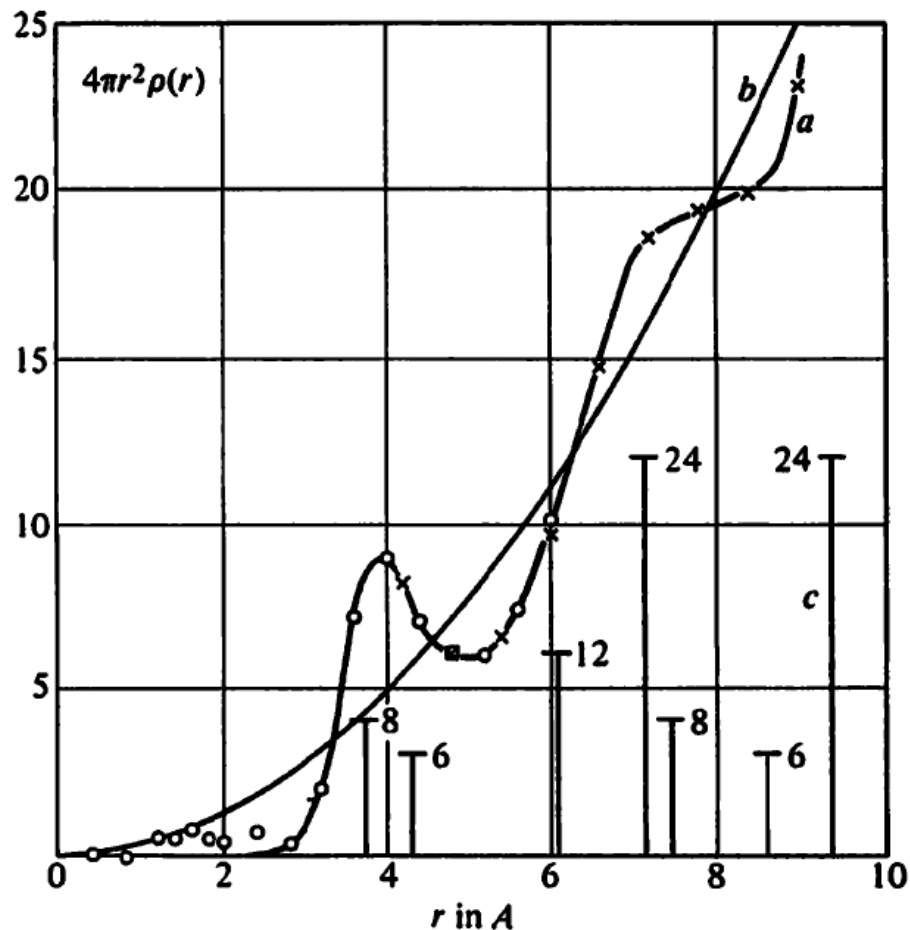


Fig. 10.6 (a) The radial distribution function $4\pi r^2 \rho(r)$ for liquid sodium. (b) The average density curve $4\pi r^2 \rho_a$. (c) The distribution of neighbors in crystalline sodium.

Thank you